## TREATMENT AND REMEDIATION PANEL

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#### **OBJECTIVES:**

- Summarize the challenges associated with treatment of PFAS-contaminated soil and groundwater
- Introduce potential viable PFAS remediation approaches

## INTRODUCTION

- Perfluorinated Compounds
  - Perfluorinated alkyl acids (PFAAs)
    - PFOA, PFOS
    - PFBA, PFBS
    - PFHxA, PFHxS
  - Intermediates or Precursors
    - N-MeFOSE
    - N-EtFOSE
    - 6:2 FTS
  - Range of properties with chain length and functional group(s) hydrophobicity, electrostatic, reactivity

Guidelines (ng/L), EPA 2016				
PFOA	70			
PFOS	70			

## INTRODUCTION



	Formula	Vapor Pressure	Aqueous Solubility	Log K <sub>oc</sub>	Degradation
PFOA	$C_8HF_{15}O_2$	0.1 kPa (20°C) 10 mm Hg (25°C)	4.1 g/L (22°C) 9.5 g/L (25°C)	2.06	Stable
PFOS	$C_8F_{17}SO_3^-$	3.31 x 10 <sup>4</sup> Pa at 20°C	570 mg/L	2.57	Stable
PFHxS	$C_6F_{13}SO_3$	0.61Pa (25°C) <sup>ES</sup>	$6.2~\mathrm{mg/L^{ES}}$ $22~\mathrm{mg/L^{ES}}$	3.5 <sup>ES</sup>	Stable
PFBS	$C_4F_9SO_3$	0.29 mm Hg at 20°C	8900 mg/L <sup>ES</sup> 344mg/L <sup>ES</sup>	2.2 <sup>ES</sup> 1.9 <sup>ES</sup>	Stable
6:2 FTS	F(CF <sub>2</sub> ) <sub>6</sub> CH <sub>2</sub> CH <sub>2</sub> SO <sub>3</sub>	0.115Pa(25°C) <sup>ES</sup> 0.00086 mm Hg (25°C) <sup>ES</sup>	11 mg/L <sup>ES</sup> 2mg/L <sup>ES</sup>	4.0 <sup>ES</sup>	Biodegradable under specific conditions

### REMEDIATION OPTIONS

- Excavation → Incineration
  - Expensive
  - Contaminants must be treated off site
- Immobilization/Stabilization
- Filtration
  - Nanofiltration
  - Reverse Osmosis
- Sorption
  - Granular Activated Carbon (GAC)
  - Carbon nanotubes
  - Biomaterials

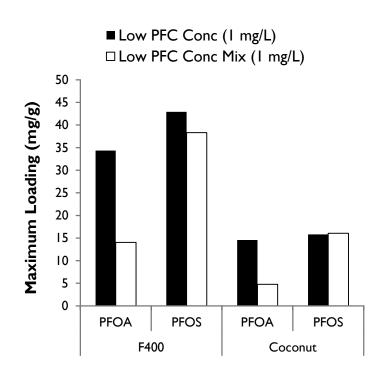
- Ion Exchange
  - Resins
  - Mineral materials (e.g., zeolites)
  - Polymers
- Precipitation/Flocculation/Coagulation
- Oxidation/Reduction
  - Chemical oxidation
  - Electrochemical, sonochemical, and photochemical
  - Plasma
  - Customized reductants

TREATMENT
TRAINS and
COMBINED
REMEDIES!

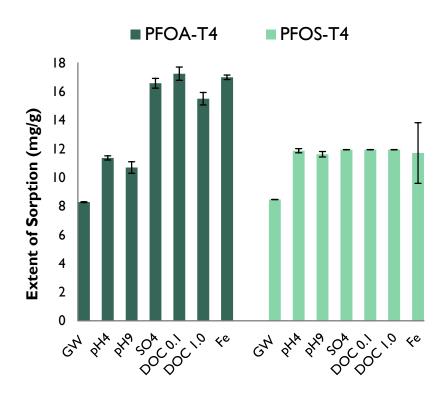
## SORPTION BY GAC



### SORPTION – GAC



■ PFOS-T4 ■ PFOA-T4 18 16 Extent of Sorption(mg/g) 14 12 10 8 No cont. Ethanol TCE TCE Kerosene No cont. Kerosene



**Carbon and PFC Type** 

Sorption under site-specific conditions...







## ION EXCHANGE

## Sustainable Removal of Poly- and Perfluorinated Alkyl Substances (PFAS) from Groundwater Using Synthetic Media





Nathan Hagelin, Amec Foster Wheeler; Steve Woodard, ECT

#### Media Selection

- Synthetic media (resins) removes various contaminants from liquids, vapor or atmospheric streams
- Isotherm testing to identify potentially effective media
- Potential for indefinite reuse via regeneration





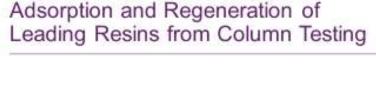


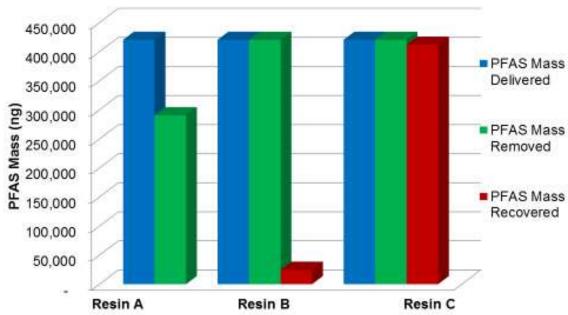








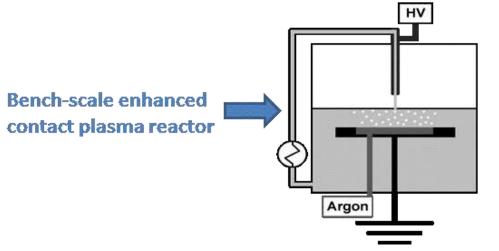




## **REDOX MANIPULATION**

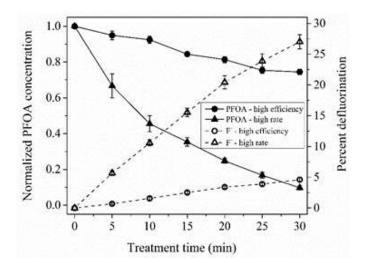
- Sonolytic
- Electrochemical
- Chemical Oxidation and Reduction
- Plasma
- Combinations...

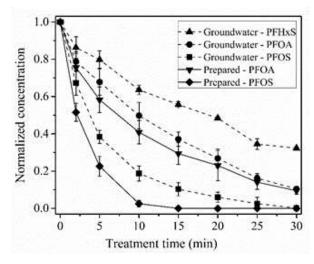
# Plasma-based water treatment: Efficient transformation of perfluoroalkyl substances (PFASs) in prepared solutions and contaminated groundwater



Plasma produces aqueous electrons and H radicals which are capable of chemically degrading PFASs







G. R. Stratton, F. Dai, C. L. Bellona, T. M. Holsen, E. R. V. Dickenson and S. Mededovic Thagard, "Plasma-based water treatment: Demonstration of efficient perfluorooctanoic acid (PFOA) degradation and identification of key reactants" Environmental Science & Technology, 2016, accepted.

Major byproducts: fluoride ions, fluorinated gases and shorter-chain PFAAs

Courtesy of Selma Mededovic Thagard, Clarkson University

## CHALLENGES AND LIMITATIONS

- Mixtures, precursors, co-contaminants
- Managing materials
  - Sorption
  - Sludge
- Incomplete mineralization
- Energy intensity
- Technical challenges to in situ treatment
- Limited field-scale examples



### **SURVEY!**

- OBJECTIVE: Improve understanding of the similarities and gaps between state of the science and the state of the practice of managing PFAS sites
  - goo.gl/zakRX3