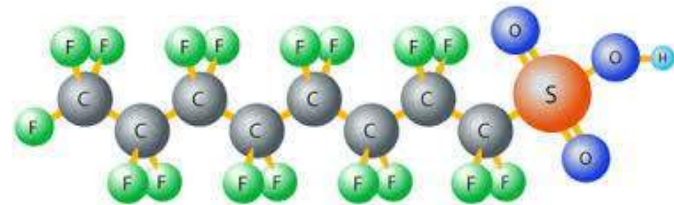




Northeast Waste Management Officials Association

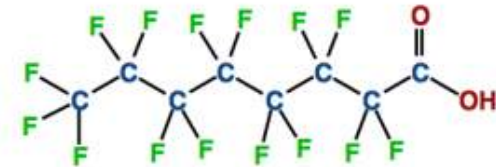
PFAS FATE & TRANSPORT



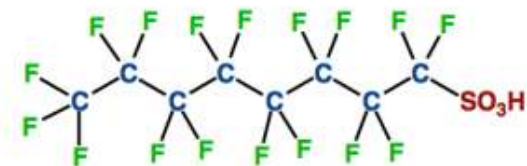
June 4, 2020

Important Concepts for PFAS Fate and Transport

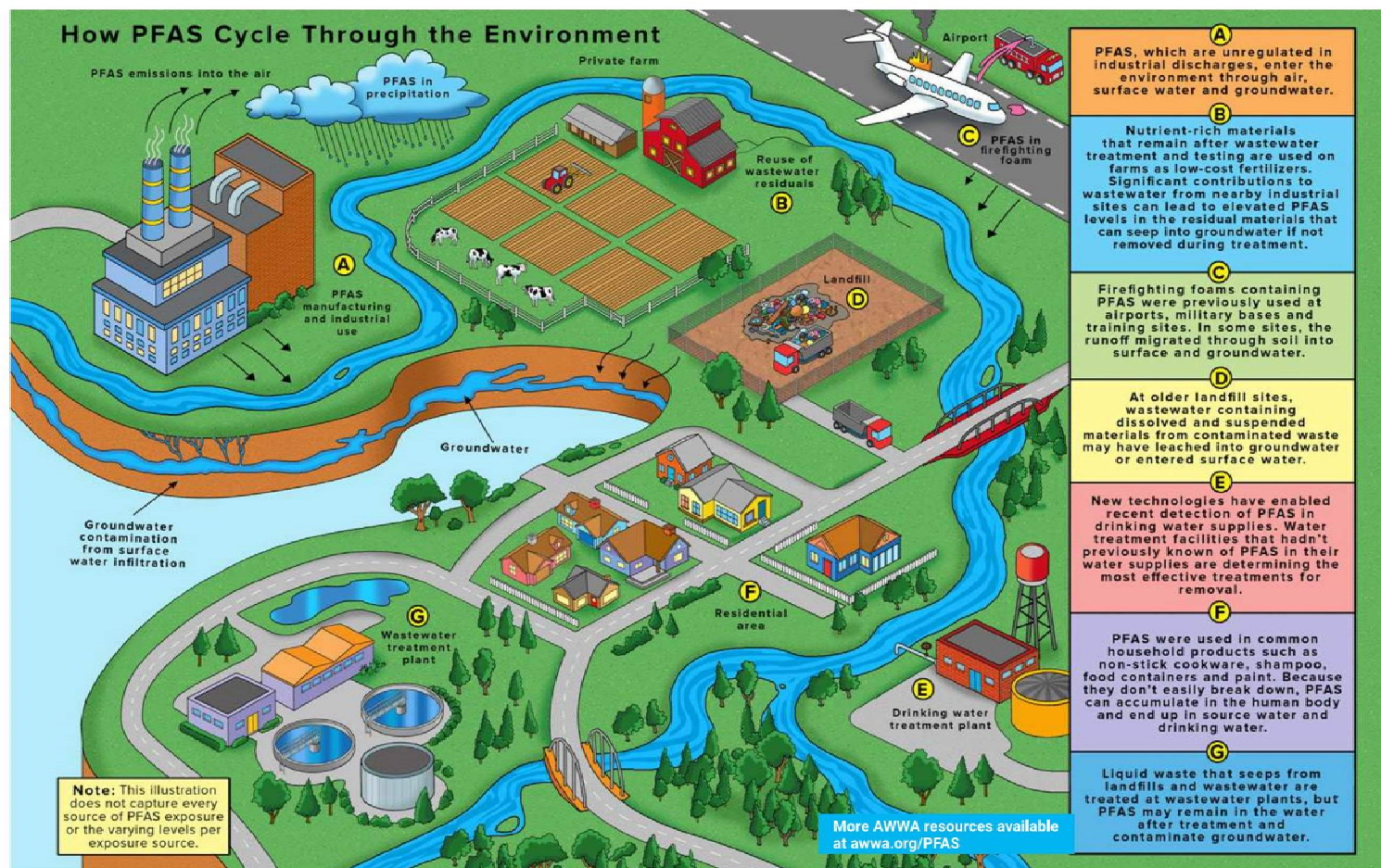
- Release Mechanisms
- Source Material Makeup
- PFAS structural make up impacts reactivity and transport (Heads and Tails)
- Persistence in Environment
- Partitioning of PFAS is complicated
 - Hydrophobic effects
 - Lipophobic effects
 - Interface interactions
 - Multiple ionic species present



PFOA - perfluorooctanoic acid

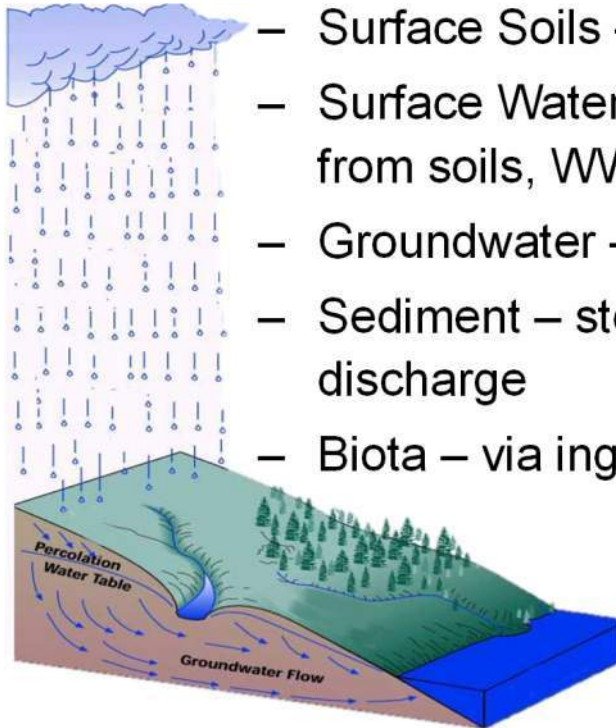


PFOS - perfluorooctanesulfonic acid



Release Mechanisms

- Use/Release of PFAS can result in impacts to:
 - Air– atmospheric transport can result in large impacted areas,
 - Surface Soils – air deposition, AFFF use, infiltration of runoff water
 - Surface Water – via direct discharge, infiltration from soils, runoff from soils, WWTF discharges
 - Groundwater – via infiltration, wastewater disposal and soil
 - Sediment – storm water infiltration, runoff of soils, groundwater discharge
 - Biota – via ingestion of impacted water, plants?, other biota



Source Type

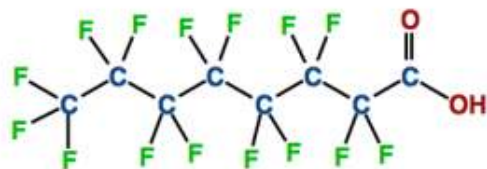
- AFFF Sources
 - AFFF is a mixture of compounds - <5% PFAS
 - There can be many PFAS (short and long) and precursors
 - Hydrocarbons from fire source
 - “Complex Mixture” in source area may effect advection, adsorption, precursor breakdown
- Manufacturing Sources
 - Can have single PFAS source or complex PFAS mixture
 - Additional compounds may be present
- Landfill Leachate
 - “Complex Mixture” in source area may effect advection, adsorption, precursor breakdown
- Wastewater Treatment Facilities
 - Multiple inputs may be present (industries, humans, surface water)
 - Treatment may cause oxidation of precursors
 - Concentration of PFAS in biosolids due to high TOC
 - Biosolids drying, composting, spreading

Structural Makeup

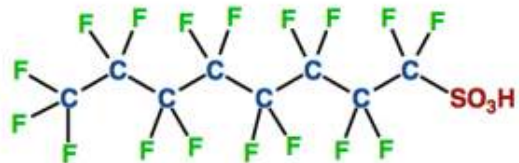
- Anionic Perfluorinated Alkyl Acids
 - Negatively charged
 - Low vapor pressure
 - Water soluble

PFAAs generally act as surfactants with tail in the air and head in water

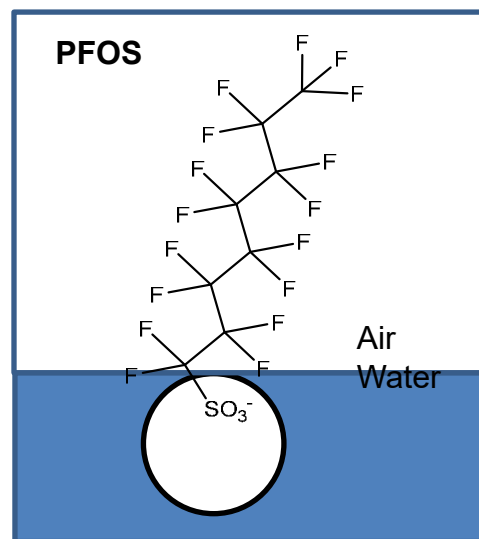
Perfluorinated TAIL Anionic HEAD



PFOA - perfluorooctanoic acid



PFOS - perfluorooctanesulfonic acid



Structural Makeup

- Anionic Perfluorinated Alkyl Acids
 - PFSAAs more strongly sorbed than PFCAs
 - Sorption generally increases with C
 - Short chains can have greater sorption than expected.
 - Retardation factors for anions can be predicted as with other contaminants (generally)

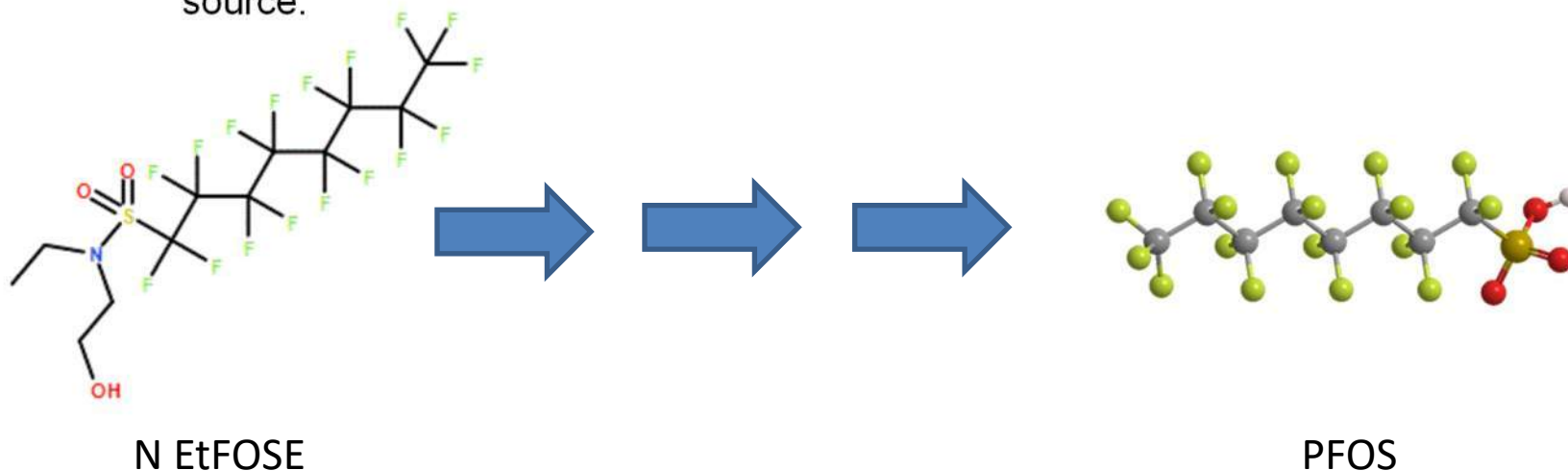
$$R_f = 1 + K_d \frac{\text{Bulk density}}{\text{Porosity}}$$

Analyte	# Carbon	K _{oc} ¹	R _f
PFBA	4	76	5
PFPeA	5	23	1.4
PFHxA	6	20	1.1
PFHpA	7	43	3
PFOA	8	78	5
PFNA	9	229	14
PFDA	10	912	57
PFUnA	11	3,600	225
PFBS	4	62	4
PFHxS	6	112	7
PFOS	8	631	39

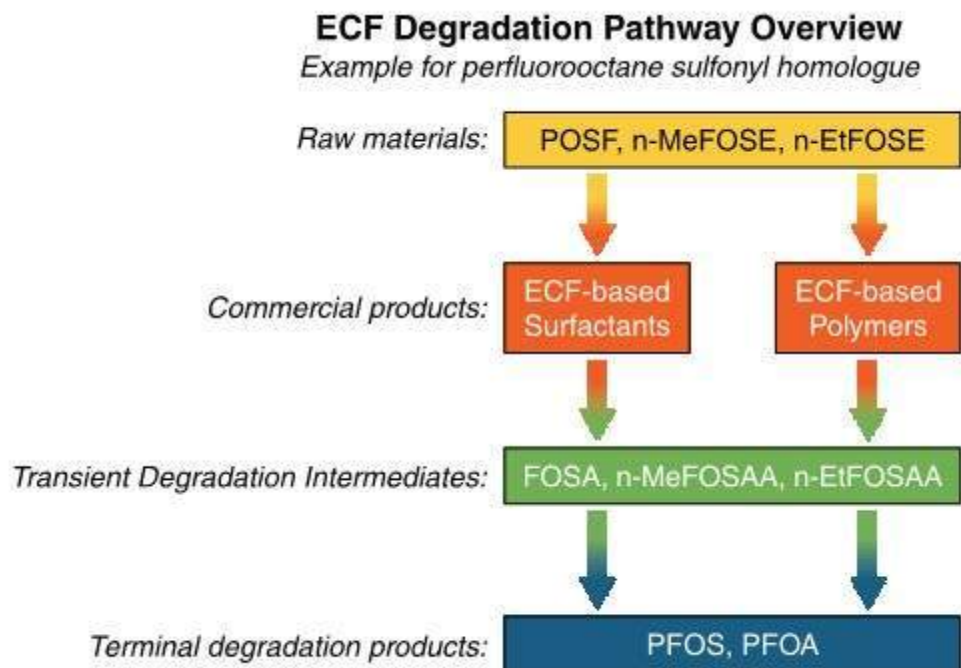
¹ Koc data from Guelfo, J.L., Higgins, C.P. Subsurface transport potential of perfluoroalkyl acids at aqueous film-forming foam (AFFF)-impacted sites. *Environ. Sci. Technol.* **2013**. 47, 4164–4171.

Structural Makeup

- Polyfluorinated Substance transport
 - State of charge may dominate retardation
 - Anions > Cations > Zwitterions
 - Short Chains generally migrate faster
 - Cation exchange onto soils may be significant....on par with organic carbon
 - Transformation into Perfluorinated end products may occur with distance from source.



Precursor Transformation

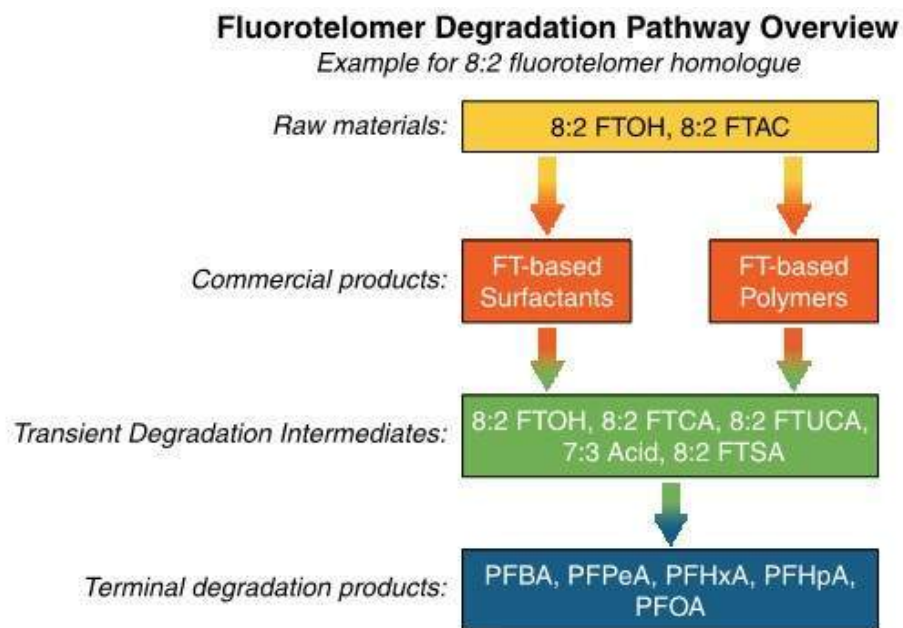


- Branched and linear isomers
- Complex mixture
- Cationic & zwitterionic
- PFSA (and some PFCAs) as end products

Figure 2-5. ECF degradation pathway overview (Example for perfluorooctane sulfonyl homologue).

Source: ITRC PFAS Fact Sheet – Fate and Transport 3/16/18

Precursor Transformation



- Built by 2's
- Fairly “clean” process yielding predictable mixtures
- PFCAs only as end products

Figure 2-4. Fluorotelomer degradation pathway overview (Example for 8:2 fluorotelomer homologue)

Source: ITRC PFAS Fact Sheet – Fate and Transport 3/16/18

What is Expected Where?

PFAS plumes have varying complexity related to source type, time since release and location relative to the source.

➤ Source Zone

- complex chemistry
- multiple “families” of contaminants
- competing transport mechanisms
- “minimal” precursor breakdown (unless remediation has occurred)

➤ Transition Zone

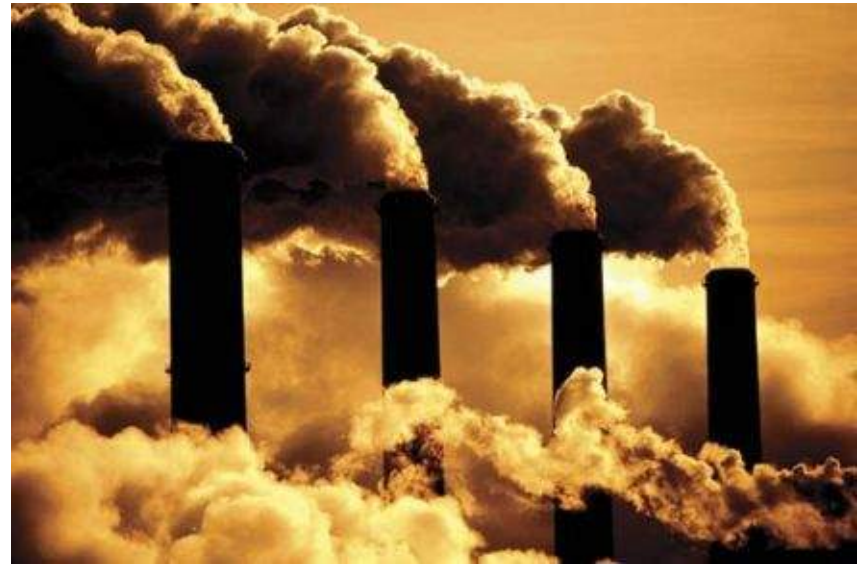
- separation of “families” via sorption/retardation/biodegradation
- precursor transformation seen (more PFAAs with distance?)
- separation of PFCAs/PFSAs and short chain/long chain via sorption/retardation

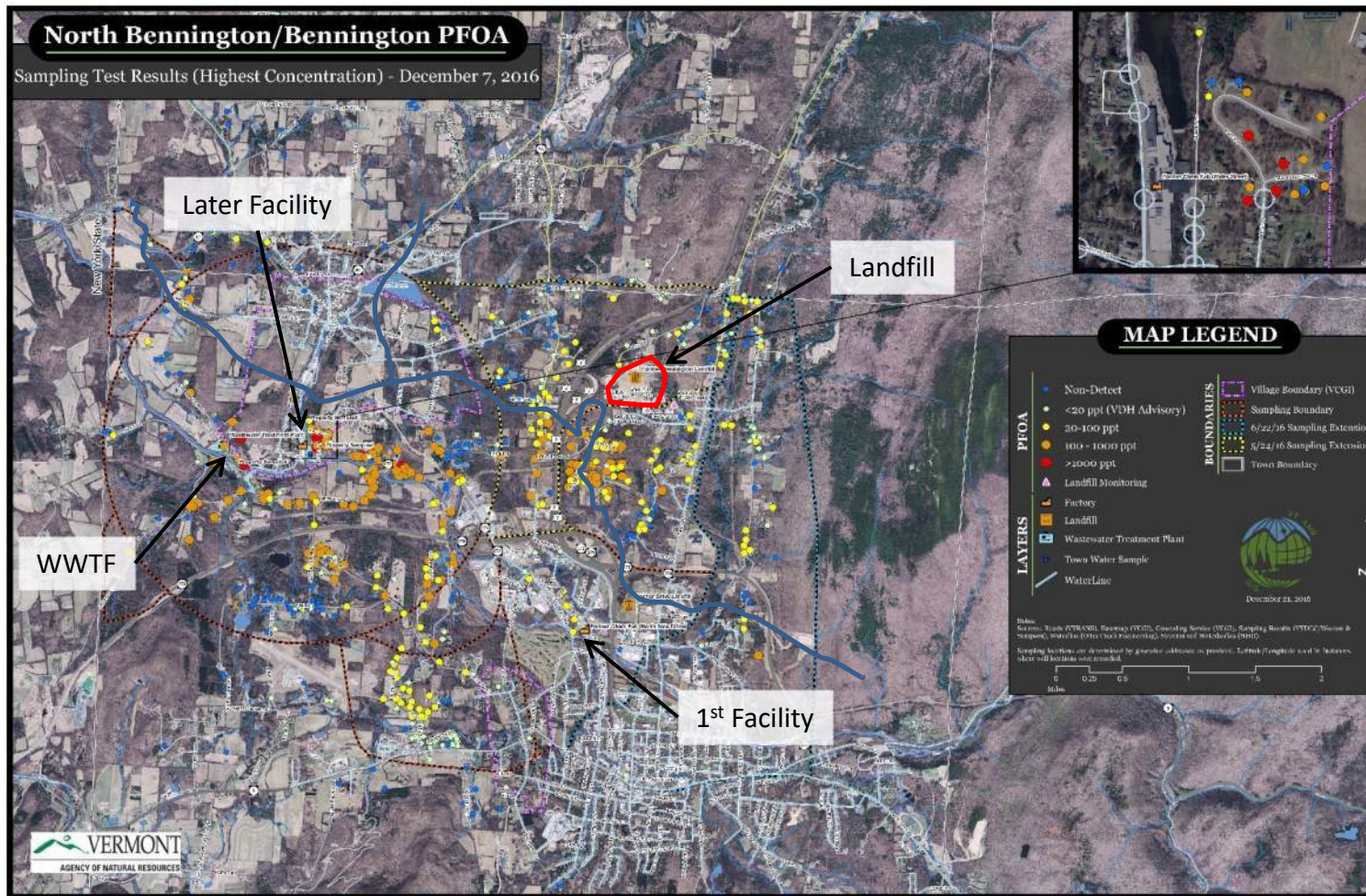
➤ Distal Zone

- “simpler” chemistry
- terminal end products dominate

Transport in Air

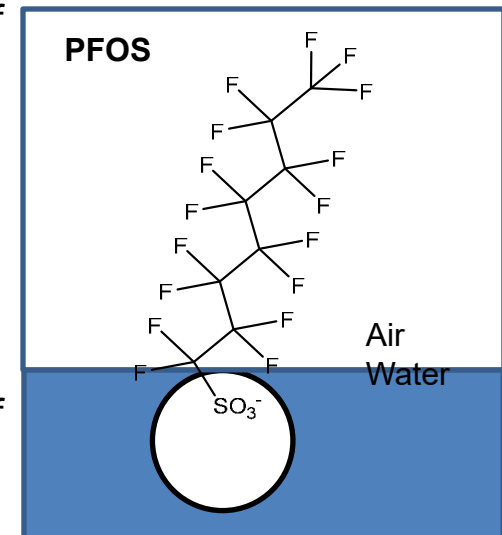
- PFAS can be volatilized by drying activities (precursor alcohols)
- “Carried” in water vapor
- “Carried” on or as particulate
- Deposition via wet and dry methods
- Transported in all wind directions, potentially miles





Transport in Vadose Zone

- Chain length and organic content of soil dependent transport dominant
- However, individual PFAS K_{oc}, ionic state and the presence of other contaminants may have significant impact
- Soil types will define advective transport via infiltration
- Adsorbed PFAS may act as “source” to groundwater for decades
- Air/Water Interface interactions likely results in retardation of migration
- Formation of micelles and interactions with NAPL



Transport in Groundwater

- Chain length and organic content of soil dependent transport dominant
- However, individual PFAS Koc, ionic state and the presence of other contaminants may have significant impact
- Soil types will define advective transport via infiltration
- Air/Water Interface interactions likely results in retardation of migration
- Formation of micelles and interactions with NAPL

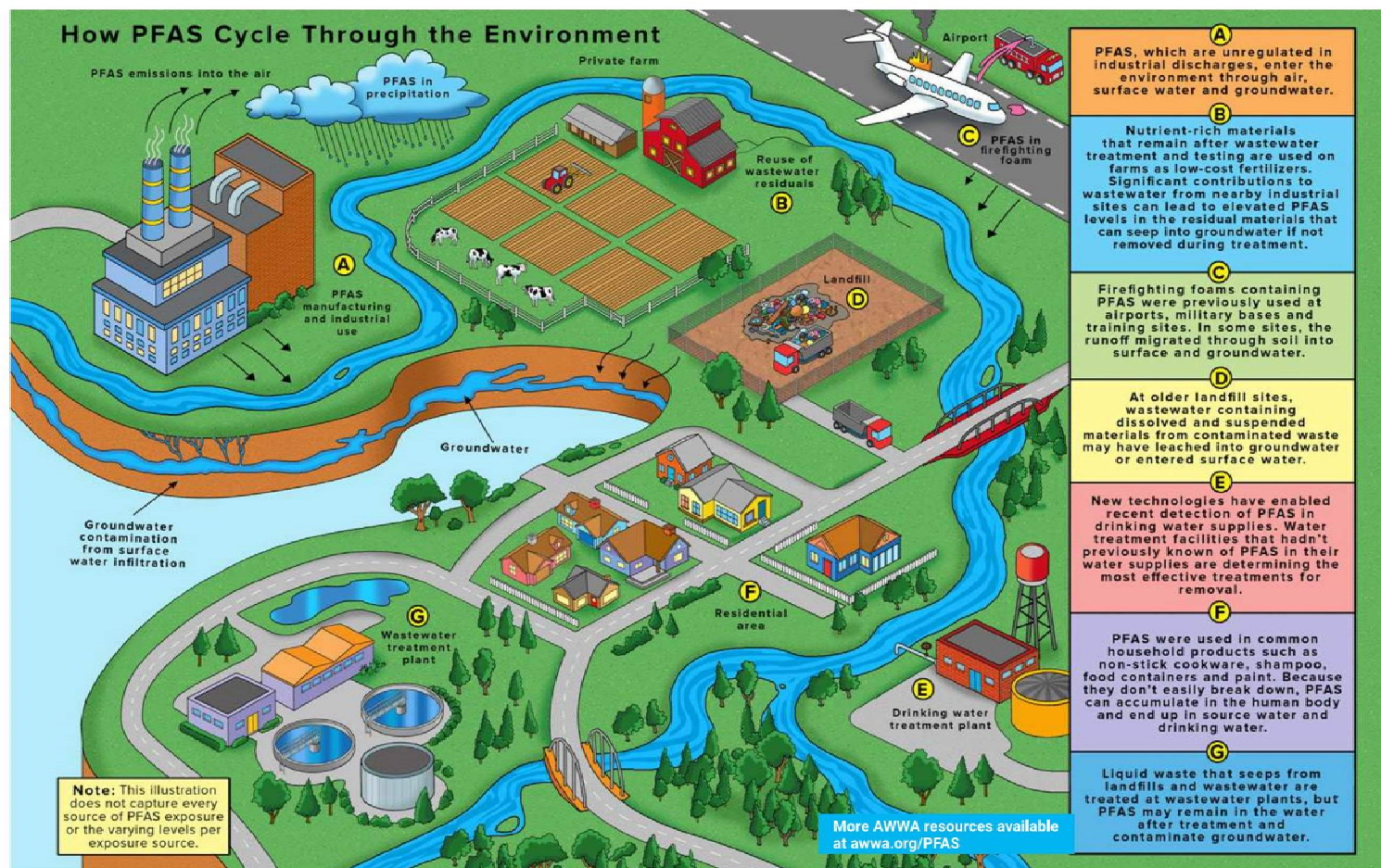


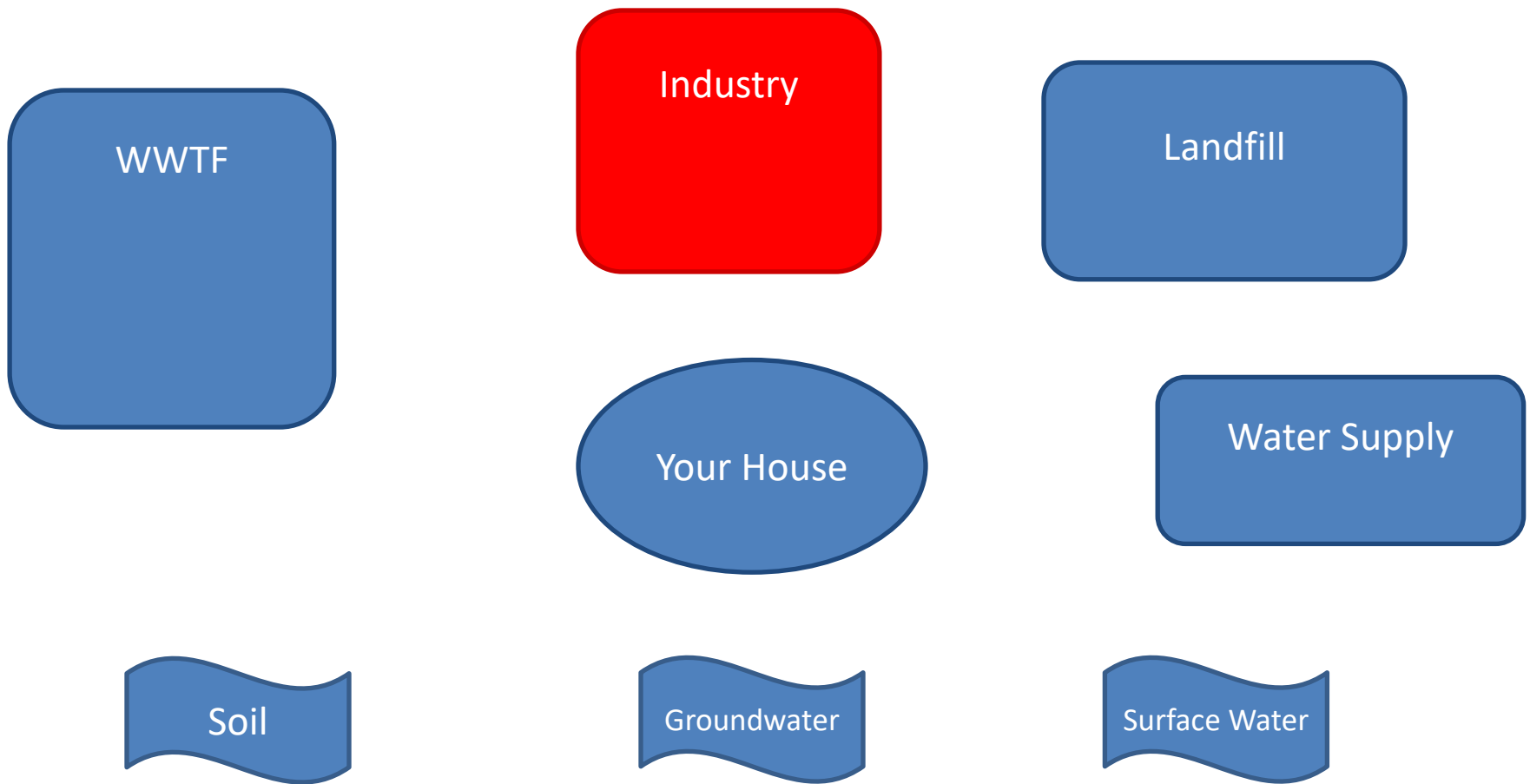
“RECYCLING”

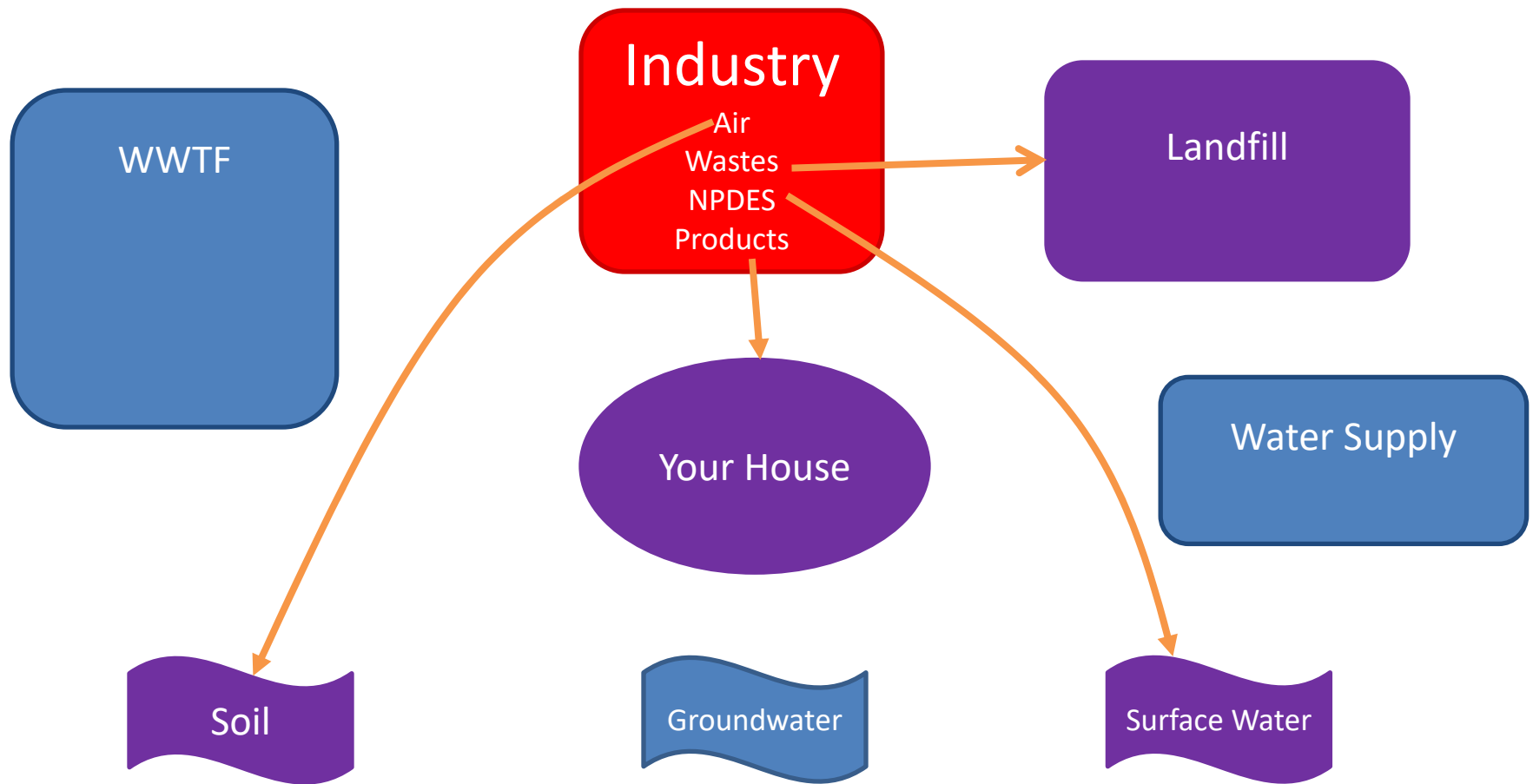
- Terminal...”FOREVER”no degradation of perfluorinated PFAS
- Soluble, “Mildly” bound to organic carbon, “Mildly” bound ionic soils.
- Nearly Ubiquitous in Environmental Media

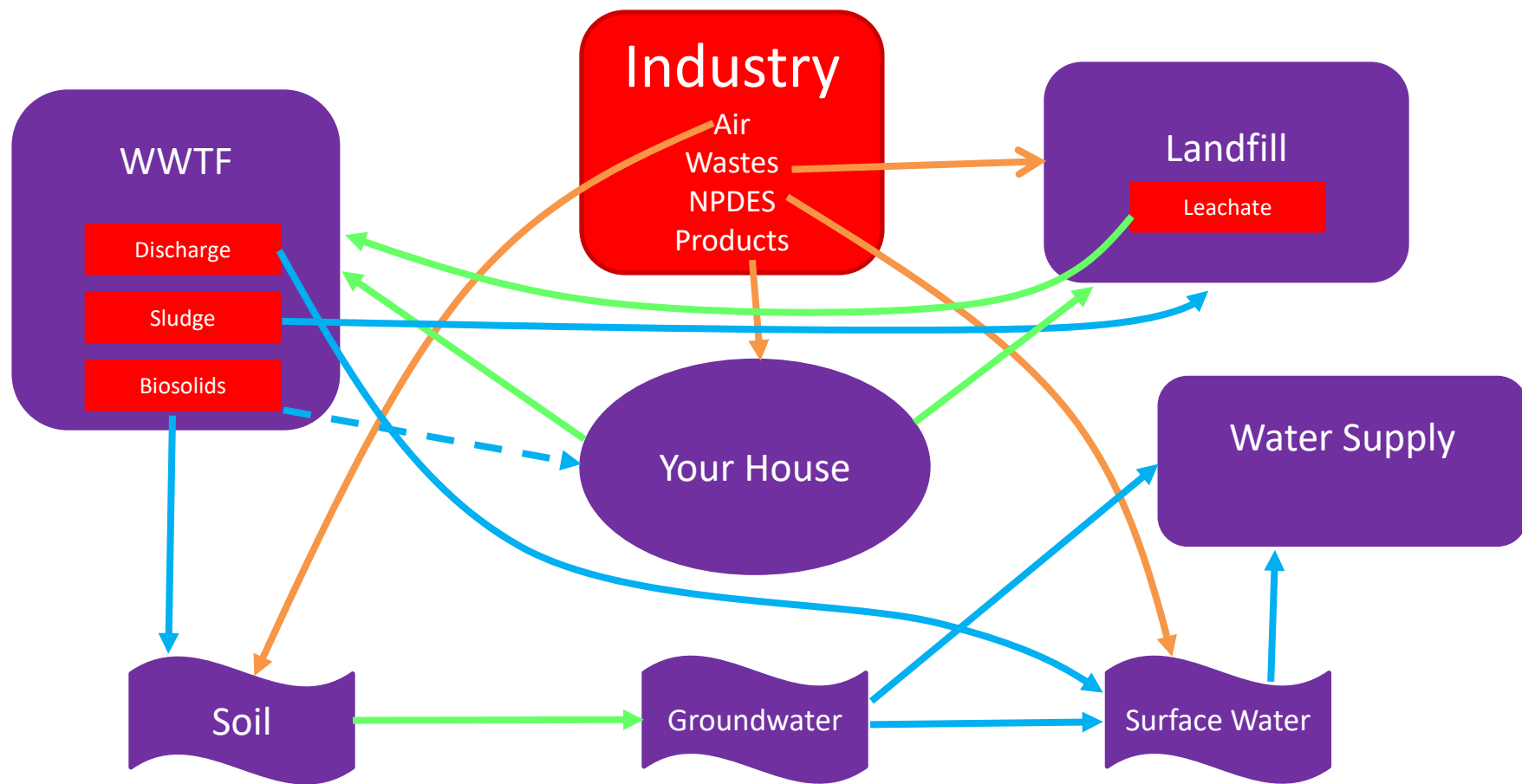
Conceptual Site Models for Release, Fate and Transport Complicated

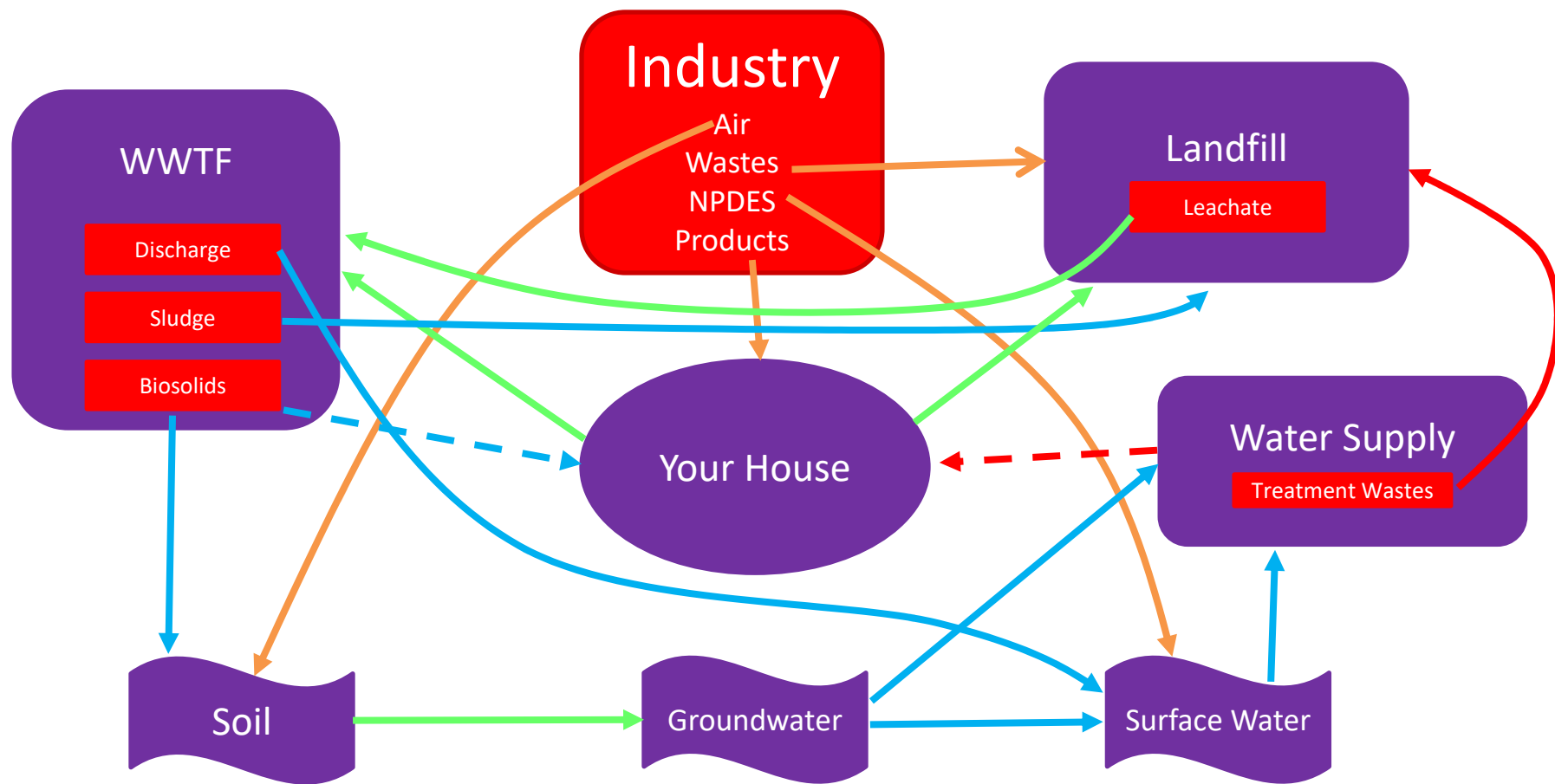


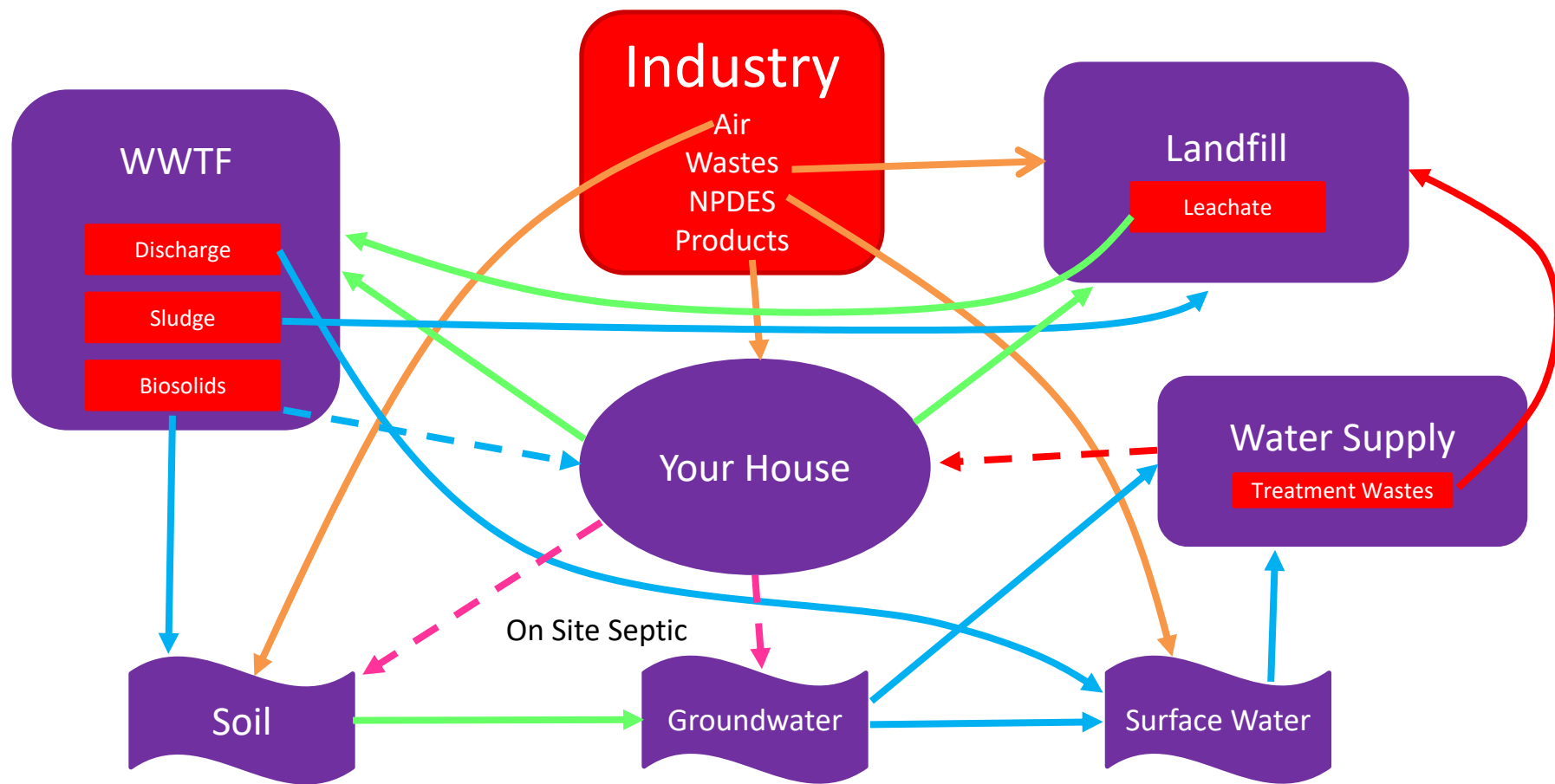












Important Concepts for PFAS Fate and Transport

- Source zones may be complex: long-term discharge potential
 - Non PFAS related contaminants can confound understanding of source area
 - Investigation areas may be larger than you're used to
- Perfluorinated PFAS:
 - TERMINAL
 - Mobility is chain-length dependent, affected by OC, pH, inorganic cations, etc.
- Polyfluorinated PFAS (precursors) are varying in their stability
 - Much more variable in terms of transport
 - Oxidizing remedial techniques (ISCO, air sparge, aerobic bioremediation) can transform precursors to TERMINAL PFAAs
- Surfactant properties may lead to increased concentrations at air/water interface or at water/NAPL interface

IMPORTANT WEBSITES

NEWMOA

<https://www.newmoa.org/>

ITRC FACT SHEETS

<https://pfas-1.itrcweb.org/fact-sheets/>

EPA PFAS Webpage

<https://www.epa.gov/pfas>

Northeastern University PFAS Project

<https://pfasproject.com/>

Questions?



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